

## Worksheet 1 Answers

1.  $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$
2.  $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$
3.  $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$
4.  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
5.  $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} + \text{NaNO}_3$
6.  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
7.  $\text{Cu} + 2\text{AgNO}_3 \rightarrow \text{Cu}(\text{NO}_3)_2 + 2\text{Ag}$
8.  $2\text{Al} + 3\text{Cu}(\text{NO}_3)_2 \rightarrow 2\text{Al}(\text{NO}_3)_3 + 3\text{Cu}$
9.  $\text{Fe} + 2\text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$
10.  $\text{Ag} + 2\text{HNO}_3 \rightarrow \text{NO}_2 + \text{H}_2\text{O} + \text{AgNO}_3$
11.  $\text{CaCO}_3 + 2\text{NaCl} \rightarrow \text{Na}_2\text{CO}_3 + \text{CaCl}_2$
12.  $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
13.  $4\text{NH}_3 + 3\text{O}_2 \rightarrow 2\text{N}_2 + 6\text{H}_2\text{O}$
14.  $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
15.  $2\text{C}_8\text{H}_{18} + 25\text{O}_2 \rightarrow 16\text{CO}_2 + 18\text{H}_2\text{O}$
16.  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
17.  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

## Worksheet 2 Answers

1.  $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$
2.  $\text{BaO} + \text{H}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + \text{H}_2\text{O}$
3.  $3\text{CaCl}_2 + 2\text{K}_3\text{PO}_4 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{KCl}$
4.  $\text{Zn} + 2\text{HCl} \rightarrow \text{H}_2 + \text{ZnCl}_2$
5.  $\text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_2 + \text{H}_2\text{S} \rightarrow \text{PbS} + 2\text{HC}_2\text{H}_3\text{O}_2$
6.  $\text{Mg} + \text{H}_2\text{SO}_4 \rightarrow \text{MgSO}_4 + \text{H}_2$
7.  $3\text{CaO} + \text{P}_2\text{O}_5 \rightarrow \text{Ca}_3(\text{PO}_4)_2$
8.  $\text{Ca} + 2\text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2 + \text{H}_2$
9.  $2\text{NaHCO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O} + 2\text{CO}_2$
10.  $\text{CuO} + \text{H}_2 \rightarrow \text{H}_2\text{O} + \text{Cu}$
11.  $\text{MgCO}_3 + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O} + \text{CO}_2$
12.  $2\text{Al}(\text{OH})_3 + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 6\text{H}_2\text{O}$
13.  $\text{AgNO}_3 + \text{HCl} \rightarrow \text{AgCl} + \text{HNO}_3$
14.  $\text{NH}_3 + 2\text{O}_2 \rightarrow \text{HNO}_3 + \text{H}_2\text{O}$
15.  $\text{ZnCl}_2 + (\text{NH}_4)_2\text{S} \rightarrow \text{ZnS} + 2\text{NH}_4\text{Cl}$
16.  $2\text{Al} + \text{Fe}_2\text{O}_3 \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$
17.  $\text{Fe} + \text{S} \rightarrow \text{FeS}$

### Worksheet 3 Answers

1.  $3\text{Zn}(\text{ClO}_3)_2 + 2\text{Al}(\text{NO}_3)_3 \rightarrow 3\text{Zn}(\text{NO}_3)_2 + 2\text{Al}(\text{ClO}_3)_3$
2.  $3\text{NaIO}_3 + (\text{NH}_4)_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + 3\text{NH}_4\text{IO}_3$
3.  $\text{K}_2\text{CrO}_4 + \text{Pb}(\text{NO}_3)_2 \rightarrow 2\text{KNO}_3 + \text{PbCrO}_4$
4.  $3\text{NiCO}_3 + 2\text{Bi}(\text{FO}_3)_3 \rightarrow 3\text{Ni}(\text{FO}_3)_2 + \text{Bi}_2(\text{CO}_3)_3$
5.  $\text{Fe}(\text{BrO}_3)_3 + 3\text{AgNO}_3 \rightarrow \text{Fe}(\text{NO}_3)_3 + 3\text{AgBrO}_3$
6.  $\text{Sb}_2(\text{SO}_4)_3 + 3\text{Ba}(\text{OH})_2 \rightarrow 2\text{Sb}(\text{OH})_3 + 3\text{BaSO}_4$
7.  $\text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_4 + 2\text{Na}_2\text{CO}_3 \rightarrow \text{Pb}(\text{CO}_3)_2 + 4\text{NaC}_2\text{H}_3\text{O}_2$
8.  $2\text{CaSO}_4 + \text{Sn}(\text{C}_2\text{O}_4)_2 \rightarrow 2\text{CaC}_2\text{O}_4 + \text{Sn}(\text{SO}_4)_2$
9.  $2\text{Li}_3\text{AsO}_4 + 3\text{Hg}_2\text{CrO}_4 \rightarrow 3\text{Li}_2\text{CrO}_4 + (\text{Hg}_2)_3(\text{AsO}_4)_2$
10.  $\text{Al}(\text{FO}_3)_3 + 3\text{LiC}_2\text{H}_3\text{O}_2 \rightarrow \text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3 + 3\text{LiFO}_3$
11.  $\text{Sn}(\text{NO}_3)_2 + \text{Mg}(\text{ClO}_3)_2 \rightarrow \text{Sn}(\text{ClO}_3)_2 + \text{Mg}(\text{NO}_3)_2$
12.  $\text{CuSO}_4 + 2\text{KOH} \rightarrow \text{K}_2\text{SO}_4 + \text{Cu}(\text{OH})_2$
13.  $3\text{FeCrO}_4 + 2(\text{NH}_4)_3\text{PO}_4 \rightarrow \text{Fe}_3(\text{PO}_4)_2 + 3(\text{NH}_4)_2\text{CrO}_4$
14.  $2\text{As}(\text{NO}_3)_3 + 3\text{Cu}_2\text{C}_2\text{O}_4 \rightarrow \text{As}_2(\text{C}_2\text{O}_4)_3 + 6\text{CuNO}_3$
15.  $2\text{NH}_4\text{OH} + \text{Hg}(\text{FO}_3)_2 \rightarrow 2\text{NH}_4\text{FO}_3 + \text{Hg}(\text{OH})_2$
16.  $\text{Sr}(\text{HSO}_4)_2 + \text{Rb}_2\text{HPO}_4 \rightarrow \text{SrHPO}_4 + 2\text{RbHSO}_4$
17.  $3\text{Mn}(\text{H}_2\text{PO}_4)_2 + 2\text{Cr}(\text{HCO}_3)_3 \rightarrow 3\text{Mn}(\text{HCO}_3)_2 + 2\text{Cr}(\text{H}_2\text{PO}_4)_3$

Worksheet 4 answers

1.  $\text{Ba}(\text{FO}_2)_2 + 2\text{NaC}_2\text{H}_3\text{O}_2 \rightarrow \text{Ba}(\text{C}_2\text{H}_3\text{O}_2)_2 + 2\text{NaFO}_2$
2.  $\text{Hg}_2(\text{NO}_2)_2 + 2\text{NH}_4\text{Cl} \rightarrow 2\text{NH}_4\text{NO}_2 + \text{Hg}_2\text{Cl}_2$
3.  $\text{Ag}_3\text{PO}_5 + \text{K}_3\text{AsO}_3 \rightarrow \text{Ag}_3\text{AsO}_3 + \text{K}_3\text{PO}_5$
4.  $2\text{LiClO} + \text{Mg}(\text{IO}_4)_2 \rightarrow 2\text{LiIO}_4 + \text{Mg}(\text{ClO})_2$
5.  $\text{PbSO}_3 + \text{Hg}(\text{ClO}_2)_2 \rightarrow \text{Pb}(\text{ClO}_2)_2 + \text{HgSO}_3$
6.  $\text{Na}_2\text{S} + \text{Cu}_2\text{CrO}_4 \rightarrow \text{Cu}_2\text{S} + \text{Na}_2\text{CrO}_4$
7.  $3\text{Ni}(\text{IO})_2 + \text{Al}_2(\text{C}_2\text{O}_4)_3 \rightarrow 2\text{Al}(\text{IO})_3 + 3\text{NiC}_2\text{O}_4$
8.  $3\text{Ba}(\text{OH})_2 + \text{As}_2(\text{SO}_3)_3 \rightarrow 3\text{BaSO}_3 + 2\text{As}(\text{OH})_3$
9.  $2\text{Li}_3\text{PO}_3 + 3\text{Zn}(\text{ClO})_2 \rightarrow 6\text{LiClO} + \text{Zn}_3(\text{PO}_3)_2$
10.  $\text{Sb}_2(\text{CO}_3)_5 + 5\text{Cu}(\text{BrO}_4)_2 \rightarrow 2\text{Sb}(\text{BrO}_4)_5 + 5\text{CuCO}_3$
11.  $\text{FeBr}_2 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{Fe}(\text{C}_2\text{H}_3\text{O}_2)_2 + \text{MgBr}_2$
12.  $3\text{Zn}(\text{NO}_2)_2 + 2\text{BiI}_3 \rightarrow 3\text{ZnI}_2 + 2\text{Bi}(\text{NO}_2)_3$
13.  $2\text{NaCl} + \text{Li}_2\text{S} \rightarrow \text{Na}_2\text{S} + 2\text{LiCl}$
14.  $\text{PbF}_2 + \text{K}_2\text{Cr}_2\text{O}_7 \rightarrow \text{PbCr}_2\text{O}_7 + 2\text{KF}$
15.  $\text{Mn}(\text{HC}_2\text{O}_4)_2 + 2\text{RbHSO}_3 \rightarrow \text{Mn}(\text{HSO}_3)_2 + 2\text{RbHC}_2\text{O}_4$
16.  $\text{Ra}(\text{HSO}_4)_2 + \text{Sr}(\text{HCO}_3)_2 \rightarrow \text{Ra}(\text{HCO}_3)_2 + \text{Sr}(\text{HSO}_4)_2$
17.  $2\text{CsOH} + \text{Be}(\text{HS})_2 \rightarrow 2\text{CsHS} + \text{Be}(\text{OH})_2$

## Worksheet 5 Answers

1.  $\text{Zn} + \text{S} \rightarrow \text{ZnS}$
2.  $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$
3.  $\text{MgBr}_2 + \text{Cl}_2 \rightarrow \text{MgCl}_2 + \text{Br}_2$
4.  $2\text{NaF} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + 2\text{HF}$
5.  $2\text{Al} + \text{Fe}_2\text{O}_3 \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$
6.  $\text{NH}_4\text{NO}_2 \rightarrow \text{N}_2 + 2\text{H}_2\text{O}$
7.  $\text{NH}_3 + 2\text{O}_2 \rightarrow \text{HNO}_3 + \text{H}_2\text{O}$
8.  $\text{Zn} + \text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{Zn}(\text{C}_2\text{H}_3\text{O}_2)_2 + \text{Pb}$
9.  $\text{CaI}_2 + \text{Br}_2 \rightarrow \text{CaBr}_2 + \text{I}_2$
10.  $\text{BaCO}_3 \rightarrow \text{BaO} + \text{CO}_2$
11.  $(\text{NH}_4)_2\text{CO}_3 \rightarrow 2\text{NH}_3 + \text{H}_2\text{O} + \text{CO}_2$
12.  $\text{Al}_2(\text{SO}_4)_3 + 3\text{Ca}(\text{OH})_2 \rightarrow 2\text{Al}(\text{OH})_3 + 3\text{CaSO}_4$
13.  $\text{C}_2\text{H}_5\text{OH} + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$
14.  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
15.  $2\text{HCl} + \text{Na}_2\text{CO}_3 \rightarrow 2\text{NaCl} + \text{H}_2\text{O} + \text{CO}_2$
16.  $\text{CO}_2 + \text{Ca}(\text{OH})_2 \rightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
17.  $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$
18.  $2\text{HCl} + \text{CuO} \rightarrow \text{CuCl}_2 + \text{H}_2\text{O}$

## Worksheet 6 Answers

1. Cu + 2AgNO<sub>3</sub> → Cu(NO<sub>3</sub>)<sub>2</sub> + 2Ag
2. 2Mg + O<sub>2</sub> → 2MgO
3. HCl + AgNO<sub>3</sub> → AgCl + HNO<sub>3</sub>
4. Mg + 2HCl → MgCl<sub>2</sub> + H<sub>2</sub>
5. Zn + 2HCl → ZnCl<sub>2</sub> + H<sub>2</sub>
6. 4Fe + 3O<sub>2</sub> → 2Fe<sub>2</sub>O<sub>3</sub>
7. Fe + S → FeS
8. Ca(OH)<sub>2</sub> + H<sub>2</sub>SO<sub>4</sub> → CaSO<sub>4</sub> + 2H<sub>2</sub>O
9. Zn + S → ZnS
10. 3Mg + N<sub>2</sub> → Mg<sub>3</sub>N<sub>2</sub>

1. CuCO<sub>3</sub> → CuO + CO<sub>2</sub>
2. 2Na + 2H<sub>2</sub>O → 2NaOH + H<sub>2</sub>
3. NH<sub>4</sub>NO<sub>2</sub> → N<sub>2</sub> + 2H<sub>2</sub>O
4. 2Cu + S → Cu<sub>2</sub>S
5. 2K + 2H<sub>2</sub>O → 2KOH + H<sub>2</sub>
6. CaCO<sub>3</sub> + 2HCl → CaCl<sub>2</sub> + H<sub>2</sub>O + CO<sub>2</sub>
7. NH<sub>4</sub>NO<sub>3</sub> → 2H<sub>2</sub>O + N<sub>2</sub>O
8. Cr + 2HCl → CrCl<sub>2</sub> + H<sub>2</sub>
9. Ba(OH)<sub>2</sub> + CO<sub>2</sub> → BaCO<sub>3</sub> + H<sub>2</sub>O